

WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁵ : C07C 43/11, C11D 1/72	A1	(11) International Publication Number: WO 94/11331 (43) International Publication Date: 26 May 1994 (26.05.94
(21) International Application Number: PCT/SE (22) International Filing Date: 12 November 1993 (30) Priority data:	(12.11.9	Stenungsund (SE). (81) Designated States: CA, FI, JP, NO, US, European paten (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU
9203478-4 19 November 1992 (19.1) (71) Applicant (for all designated States except US): NOBEL AB [SE/SE]; S-444 85 Stenungsund ((72) Inventors; and (75) Inventors/Applicants (for US only): DAHLGREN [SE/SE]; Mjölnarvägen 4, S-444 95 Ödsn BERGSTRÖM, Karin [SE/SE]; PL 3842, S-44; gälv (SE).	: BER((SE). N, Lenn	Published With international search report. art E).
	·	H
(54) Title: USE OF ALKOXYLATE OF 2-PROPYL	. НЕРТ	ANOL

(57) Abstract

The invention relates to the use of an alkoxylate having the general formula (I): C₅H₁₁CH(C₃H₇)CH₂O(A)_nH, wherein A is an alkyleneoxy group having 2-4 carbon atoms and n is 2-16, preferably 3-12, in detergent compositions for degreasing hard surfaces.

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
ĀŪ	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinca	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Konya	RO	Romania
CA	Canada	KG	Kyrgystan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic	SD	Sudan
CG	Congo		of Korca	SE	Sweden
CH	Switzerland	KR	Republic of Korea	SI	Slovenia
CI	Côte d'Ivoire	KZ	Kazakhstan	SK	Slovakia
CM	Cameroon	니	Liechtenstein	SN	Senegal
CN	China	LK	Sri Lanka	TD	Chad
CS	Czechoslovakia	LU	Luxembourg	TG	Togo
cz	Czech Republic	LV	Latvia	TJ	Tajikistan
DE	Germany	MC	Monaco	TT	Trinidad and Tobago
DK	Denmark	MD	Republic of Moldova	UA	Ukraine
ES	Spain	MG	Madagascar	US	United States of America
FI	Finland	ML	Mali	UZ	Uzbekistan
FR	France	MN	Mongolia	VN	Vict Nam
GA	Gabon				
	Gabon				

1

USE OF ALKOXYLATE OF 2-PROPYL HEPTANOL

The present invention relates to the use of an alkoxylate of 2-propyl heptanol in compositions for cleaning hard surfaces. The alkoxylate shows low foaming compared with similar compounds having a hydrophobic group of the same size. The alkoxylate may advantageously be used as a surface-active component in detergent compositions.

10 It has long been known to alkoxylate alcohols for obtaining non-ionic surface-active compounds. These compounds have been used e.g. in detergent compositions because of their wetting and dispersing properties. In a number of applications, alkoxylates of C₈₋₁₁ alcohols 15 have however been found to be too high-foaming and/or not to have the desired detergent power. For example, ethoxylates based on branched C₈ alcohols often exhibit acceptable foaming but too low a detergent power, whereas ethoxylates based on straight or branched alcohols having 20 a larger hydrocarbon chain often show an acceptable surface activity but too high foaming. Thus, there is a need for new alkylene oxide adducts with an improved ratio of foaming to detergent power.

It has now been found that an alkoxylate based on 2-propyl heptanol is suitable for use as a detergent in compositions for cleaning hard surfaces, since it has good detergent and wetting properties as well as low foaming as compared with other alcohols having substantially the same chain length. In formulations, the alkoxylate also has a desirable thickening effect. In addition, it has been found that the alkoxylate is easily degradable and has a surprisingly low biotoxicity. In tests, no skin-irritant effect has been noted.

The alkoxylate for use according to the invention can 35 be illustrated by the formula

 $C_5H_{11}CH(C_3H_7)CH_2O(A)_nH$ (I)

wherein A is an alkyleneoxy group having 2-4 carbon atoms and n is 2-16, preferably 3-12. Preferably, 50-100% of all alkyleneoxy groups are ethyleneoxy groups. In those cases where different alkyleneoxy groups are present in the same compound, they may be added randomly or in block.

Generally, the alkoxylate is an ethoxylate having 3-7, preferably 4-6 ethyleneoxy groups.

In an advantageous mode of execution, ethylene oxide can be added in a first step and thereafter alkylene oxide having 3-4 carbon atoms. These compounds can be illustrated by the formula

 $_{15}$ $_{5}^{H_{11}CH(C_{3}^{H_{7}})CH_{2}^{O(C_{2}^{H_{4}^{O}})_{p}(B)_{r}^{H}}$ (II)

20

25

30

35

wherein B is an alkyleneoxy group having 3-4 carbon atoms, p is 1-10 and r is 1-6. Preferably, p is 2-8 and r is 1-4. These compounds have lower foaming than the corresponding compounds without any alkyleneoxy groups having 3-4 carbon atoms.

The alkoxylates for use according to the invention described above can be prepared by adding in a conventional manner in the presence of a conventional alkali catalyst, such as potassium hydroxide or sodium hydroxide, the above-mentioned amounts of alkylene oxide to 2-propyl heptanol, which is a so-called Guebert alcohol. According to a preferred mode of execution, the addition of ethylene oxide is performed using a conventional catalyst which gives a narrower distribution of added ethylene oxide than any alkali catalyst, such as NaOH or KOH. Thus prepared alkoxylates according to the invention have very low foaming. Examples of conventional catalysts giving a narrow distribution of added alkylene oxide are Ca(OH)2, Ba(OH)2, Sr(OH)2 and hydrotalcite. The reaction is preferably conducted in the absence of free water to reduce the amount of by-products and usually at a temperature of 70-180°C.

3

The alkoxylate, especially ethoxylate and alkoxylate of formula II, is suitably included in compositions for cleaning hard surfaces, e.g. for degreasing or dishwashing. Especially good results have been obtained when degreasing lacquered or unlacquered metal surfaces. In addition to the alkoxylate, the compositions may also contain other detergents, such as anionic surface-active compounds. Examples hereof are alkyl sulphate, alkyl ether sulphate, alkyl benzene sulphonate, α-olefin sulphonate and alkyl glyceryl sulphonate. Also, the compositions may contain solutising additives, complexing agents and/or pH-adjusting agents, enzymes, other surface-active components, bactericides and perfumes. The compositions are usually aqueous and in the form of emulsions, microemulsions or solutions.

The invention will be further illustrated by the following Examples.

Example 1

20 Alkoxylates according to the invention are prepared by alkoxylating 2-propyl ethanol with the amounts of alkylene oxide appearing from the Table below in the presence of potassium hydroxide or Ca(OH)₂ as catalyst. The resulting products were analysed and structurally determined by gas chromatography and mass spectrometry. The turbidity points were measured in water or monobutylether diethylene glycol. The following results were obtained.

Table 1

Com- pound	Alcohol	Mole of alkylene oxide/mole	Cata- lyst	Turbidi point	lty
	·	of alcohol		Water	BDG
					40
1	2-propyl heptanol	3.0 EO	кон	-	40
2	2-propyl heptanol	5.5 EO	кон	-	62
3	2-propyl heptanol	8.4 EO	кон	60	73
4	2-propyl heptanol	3.0 EO	Ca(OH)2	-	29
5	2-propyl heptanol	5.0 EO	Ca(OH)	-	52
6	2-propyl heptanol	7.0 EO	Ca(OH)2	-	61
A	2-ethyl hexanol	2 EO	КОН	-	28
В	2-ethyl hexanol	5 EO	кон	42	-
С	C ₉₋₁₁ alcohol ²⁾	4 EO	кон	-	62
D	Collin alcohol2;	6 EO	кон	56	-
E	Co_11 alcohol2,	8 EO	кон	78	-
F	C ₁₁ alcohol ³	. 3 EO	кон	-	51
G	C ₁ , alcohol ³)	5 EO	кон	27	-
н	C ₉₋₁₁ alcohol ²⁾	4 EO	Ca(OH)2	· –	57

EO = ethylene oxide; PO = propylene oxide,

BDG = monobutylether diethylene glycol

Example 2

The foaming properties of the alkoxylates reported in the following Table were measured according to Ross-Miles ASTM D 1173-53. The following results were obtained.

¹⁾ PO added first 2) Dobanol 91 from Shell

³⁾ Dobanol 1 from Shell

Table 2

	Compound	Foam height, cm		
5		O min	5 min	
	2	18	7	
	4	0	0	
•	5	5	0	
10	6	10	5	
10	A	40	10	
٠	В	50	0	
	С	80	20	
	D	95	30	
15	E	45	15	
	н	20	5	

From these results it appears that the compounds according to the invention have lower foaming than the most closely related reference products. Thus, compound 2 has lower foaming than compounds A, B, C, D and E, while compounds 4, 5 and 6 have lower foaming than all the reference compounds.

Example 3

On a vertically disposed, lacquered iron plate, smeared with mineral oils, soot, salts and clay was applied 20 ml of a detergent composition made up of the following components.

$\overline{}$	\sim
3	u
_	_

35

25

Parts by weight	Component
34	Compound 6 or H
67	NTA
27	Ethoxylated quaternary fatty amine
20,000	Water

6

The effect achieved was evaluated both with respect to the area of the cleaned surface (i.e. wettability) and with respect to the cleanness of the cleaned surface. Cleanness was evaluated according to an ascending scale of 1-10, where 1 indicates that no improvement of the cleanness could be observed and 10 indicates a completely clean surface. The following results were obtained.

Table 3

10

Compound	Cleaned surface, cm	Cleanness
6	95	9
н	54	8

15

From these results it appears that the formulation containing the compound according to the invention yielded both improved cleanness and a larger cleaned area.

20 Example 4

The microtoxicity, which is a measure of the aquatic toxicity, was determined for the compounds below as the water concentration of the compounds at which the ability of luminescent bacteria to emit light for 5 min is reduced by 50%. The following results were obtained.

Table 4

2	n	

35

Compound	Concentrate, ppm
2	42
5	31
С	2
D	3
F	1
G	2

From these results it appears that the compounds according to the invention have essentially lower microtoxicity than the reference compounds.

8

CLAIMS

wherein A is an alkyleneoxy group having 2-4 carbon atoms and n is 2-16, preferably 3-12, in a detergent composition for hard surfaces.

- 2. Use as claimed in claim 1, characterised in that 50-100% of all the alkyleneoxy groups are ethyleneoxy groups.
- 3. Use as claimed in claim 1, character is ed in that A is ethyleneoxy groups and n is 3-7.
 - 4. Use as claimed in claim 1 or 2, characterised by having the formula
- 20 $C_5H_{11}CH(C_3H_7)CH_2O(C_2H_4O)_p(B)_rH$ (II) wherein B is an alkyleneoxy group having 3-4 carbon atoms, p is 1-10 and r is 1-6, p preferably being 2-8 and r preferably being 1-4.
- 5. Use as claimed in any one of claims 1-4, c h a r a c t e r i s e d in that the detergent composition is a composition for degreasing unlacquered or lacquered metal surfaces.
- 30 6. An alkoxylate, characterised by having the formula

$$C_5H_{11}CH(C_3H_7)CH_2O(C_2H_4O)_p(B)_rH$$
 (II)

wherein B is an alkyleneoxy group having 3-4 carbon atoms, 35 p is 1-10 and r is 1-6, p preferably being 2-8 and r preferably being 1-4.

A. CLASSIFICATION OF SUBJECT MATTER

IPC5: C07C 43/11, C11D 1/72
According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC5: C07C, C11D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

SE,DK,FI,NO classes as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

CA

5, A, 3862243 (THOMAS J. BELLOS), 21 January 1975 (21.01.75)	1-6
	·
	1
S, A, 3340309 (EUGENE A. WEIPERT), 5 Sept 1967 (05.09.67)	1-6
	
5, A, 2508036 (MILTON KOSMIN), 19 November 1947 (19.11.47)	1-6
·	
S, A, 3567784 (WILLIAM T. TSATSOS ET AL.), 2 March 1971 (02.03.71)	1-6
	,
	, A, 2508036 (MILTON KOSMIN), 19 November 1947 (19.11.47) , A, 3567784 (WILLIAM T. TSATSOS ET AL.),

X.	Further documents are listed in the continuation of Box	· C.	X See patent lanning annox.
*	Special categories of cited documents:	T"	later document published after the international filing date or priority date and not in conflict with the application but cited to understand
A	document defining the general state of the art which is not considered to be of particular relevance		the principle or theory underlying the invention
"E"	erier document but published on or after the international filing date	"X"	document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive
"L"	document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other		step when the document is taken alone
	special reason (as specified)	"Y"	
"O"	document referring to an oral disclosure, use, exhibition or other means		considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"P"	document published prior to the international filing date but later than the priority date claimed	*&*	
Date	e of the actual completion of the international search	Date	of mailing of the international search report
			2 5 -02- 1994
22	February 1994		
Nan	ne and mailing address of the ISA/	Autho	rized officer
	edish Patent Office		
	5055, S-102 42 STOCKHOLM	Eva	Johansson
	simile No. +46 8 666 02 86	Telepi	none No. +46 8 782 25 00
	PCT/IS A (210 (second sheet) (July 1992)		

Form PCT/ISA/210 (second sheet) (July 1992)



International application No.
PCT/SE 93/00967

		PC1/3E 93/00	
C (Continu	ation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the rele	evant passages	Relevant to claim No.
A	EP, A1, 0046582 (CONOCO INC.), 3 March 1982 (03.03.82)		1-6
	(03.03.62)		
		•	
	·		
	Constitution (Indiana)		<u> </u>

Form PCT/ISA/210 (continuation of second sheet) (July 1992)



Interional application No.
PCT/SE 93/00967

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
IS-A-	3862243	21/01/75	NONE		
US-A-	3340309	05/09/67	BE-A-	683227	01/12/66
			DE-A-	1593043	23/07/70
			FR-A-	1484951	00/00/00
			GB-A-	1131333	00/00/00
			NL-C-	134504	00/00/00
			NL-A-	6608854	29/12/66
IS-A-	2508036	19/11/47	NONE		
IS-A-	3567784	02/03/71	NONE		
EP-A1-	0046582	03/03/82	SE-T3-	0046582	
			CA-A-	1157052	15/11/83
			JP-C-	1555089	23/04/90
			JP-A-	57042646	10/03/82
			US-A-	4302613	24/11/81

